

THE TOTAL SYNTHESIS OF 1-OXYGENATED EUDESMANE SESQUITERPENES :

( $\pm$ ) DIHYDROREYNOSIN AND ( $\pm$ ) 1-OXOCOSTIC ACID

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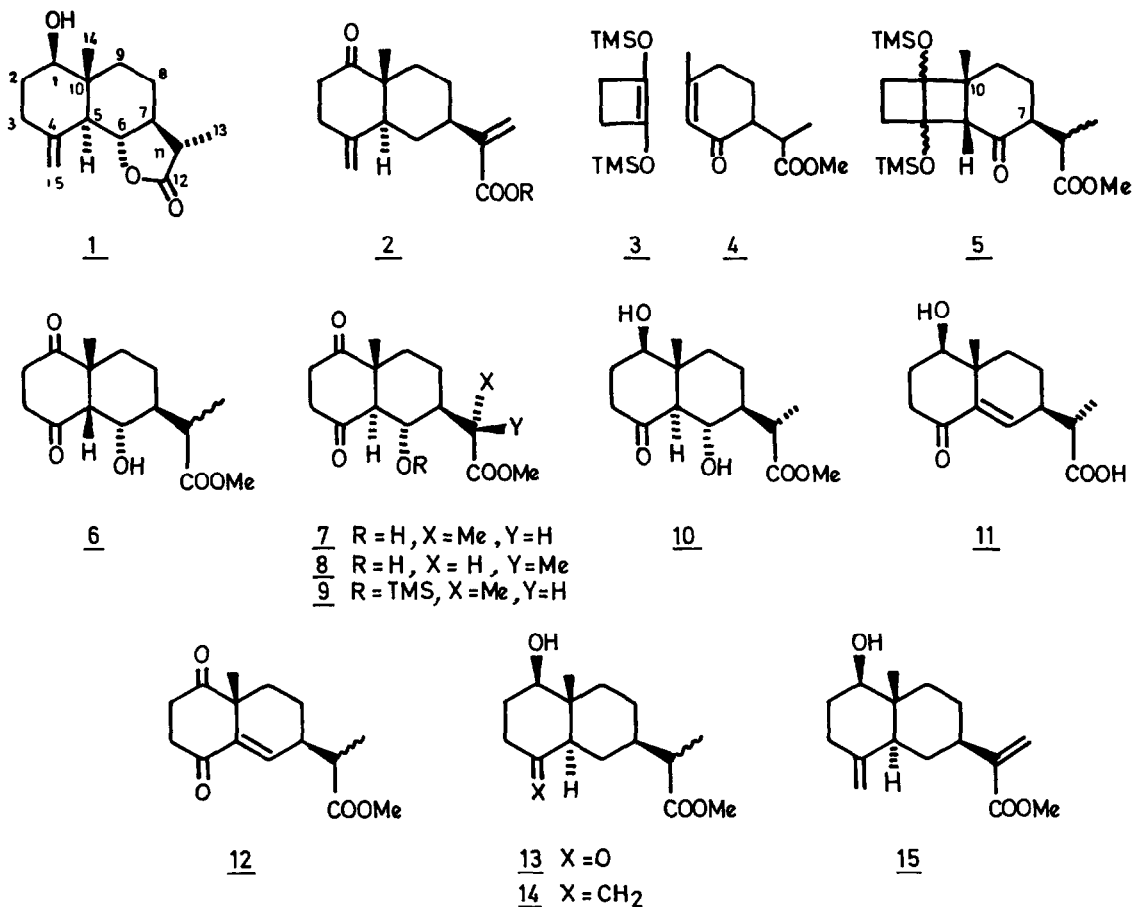
SUMMARY

The total synthesis of ( $\pm$ ) dihydroreynosin 1 and of ( $\pm$ ) 1-oxocostic acid 2 (R = H) are illustrative examples for a short entry into 1-oxygenated eudesmanes via an easily accessible key intermediate 7.

Many total syntheses of eudesmanes have been accomplished<sup>2</sup>; however the 1-oxygenated numbers comprise an important subclass which has received little attention from a synthetic viewpoint<sup>3</sup>. We want to describe a short highly stereo- and regioselective approach to this subclass with special emphasis on those members which carry a heavily functionalized C-7 "isopropyl group" amenable for lactone ring formation; therefore including also some eudesmanolides as target molecules. Eudesmanolides<sup>4</sup> possess a  $\gamma$ -lactone ring mostly closed in a trans manner toward C-6 and about 45 % of them carry an oxygen function at C-1. Many members contain 3,4-, 4,5-, 4,15, 11,13-double bonds. A characteristic feature of the eudesmane framework is the relative cis configuration of the C-7 side chain and the angular methyl group. The approach is based on previously discussed photocycloaddition reactions of 1,2-bis(trimethylsiloxy)cycloalkenes<sup>5</sup> with 2-cycloalkenones and allows a facile construction of key intermediate 7 (+8). The potentiality of 7 (+8) is illustrated by the synthesis of ( $\pm$ ) dihydroreynosin 1<sup>6</sup> and of ( $\pm$ ) 1-oxocostic acid 2<sup>7</sup> (R=H).

All, but one, of the carbon atoms of the eudesmane skeleton were assembled during the photocycloaddition of 3 and 4 (350 nm, pentane, N<sub>2</sub>, r.t.). The crude product (5; major isomer<sup>8</sup>) was reduced (NaBH<sub>4</sub>, MeOH, -20°C, 1 h); subsequent hydrolysis and oxidative cleavage (NaIO<sub>4</sub>, H<sub>2</sub>O-MeOH, r.t., 30 min, dark) afforded 6 which upon column chromatography on SiO<sub>2</sub> partially isomerized to the trans fused epimers 7 and 8 (36 % combined overall yield from 4). Stirring an ether solution of 6 in the presence of SiO<sub>2</sub> (r.t., 48 h) cleanly led to an equilibrium mixture (7 + 8 : 6 = 9:1) from which the epimers 7 and 8 were isolated in 87 % yield upon crystallization from ether. Epimers 7 and 8 will be potential key intermediates if both carbonyl functions can be differentiated. During the study for this differentiation a remarkable observation was made. Starting from the epimers 7 and 8, a complete stereo-

homogeneous TMS ether 9 was formed (98 % yield;  $\text{TMSCl}$ ,  $\text{Et}_3\text{N}$ , DMF, r.t., 2 h) as was proven by the  $^1\text{H}$  NMR spectrum (9 :  $13\text{-CH}_3$ ;  $\delta = 1.10$ ,  $J = 7$  Hz, 1 doublet; 7+8 : 2 doublets at 1.13 and 1.20 ppm ( $J = 7.25$  Hz)). Concomitant equilibration to the most thermodynamically stable C-11 isomer had occurred; this was confirmed by treatment of the mixture 7 and 8 with  $\text{Et}_3\text{N}$  in DMF (2 h at r.t.) which led to pure 7 |m.p.  $143^\circ\text{C}$ ;  $\nu$  1735, 1710  $\text{cm}^{-1}$ ;  $\delta$  ( $\text{CDCl}_3$ ) 3.93 1H, ddd, 9.50 Hz, 11 Hz and 2.25 Hz), 3.68 (3H, s), 3.35 (OH, 1H, d, 2.25 Hz), 1.13 (3H, d, 7 Hz), 1.03 (3H, s);  $m/z$  at 282 ( $\text{M}^+$ , 3) 195 (100)|.



This result suggests a hydrogen bond between the ester and hydroxyl functions, thereby fixing the side chain in a preferred cyclic conformation. As can be deduced from Dreiding models the most stable configuration should carry an  $\alpha$  oriented  $13\text{-CH}_3$  group (vide infra).

Formation of the TMS ether 9 not only provided a highly stereoselective route but also allowed complete regiocontrol<sup>9</sup>. The TMS ether function efficiently blocked the 4-carbonyl group;  $\text{NaBH}_4$  reduction of 9 in MeOH ( $-30^\circ\text{C}$ , 10 min) and subsequent acid hydrolysis afforded stereohomogeneous 10 in 95 % yield

after crystallization from ether |10 : m.p. 153-154°C;  $\nu$  : 3420, 1730, 1710  $\text{cm}^{-1}$ ,  $\delta$  ( $\text{CDCl}_3$ ) 3.86 (1H, ddd, 9.5, 9, 2.5 Hz), 3.84 (1H, dd, 11.5, 4.5 Hz), 3.68 (3H, s), 3.1 (OH; 1H, d, 2.5 Hz), 2.97 (1H, qd, 7.25 Hz, 4.75 Hz), 2.22 (1H, d, 9.5 Hz), 1.12 (3H, d, 7.25 Hz), 0.8 (3H, s); m/z at 284 ( $\text{M}^+$ , 1), 41 (100)|. Ketone 10 is a direct precursor for the synthesis of ( $\pm$ ) dihydroreynosin 1; formation of 1 proved the assumed configuration at C-11 in 7. Treatment of 10 with an excess methylene triphenylphosphorane (THF, HMPA, 1:3) gave albert in low yield (21 %) directly ( $\pm$ ) dihydroreynosin 1; the major product was keto acid 11 (71 %) occurring from  $\beta$ -elimination. The spectral properties of synthetic 1 |m.p. 108°C;  $\nu$  : 3360, 1747  $\text{cm}^{-1}$ ,  $\delta$  ( $\text{CDCl}_3$ ) 4.97 (1H, s), 4.83 (1H, s), 4.05 (1H, dd, 10.5 and 10.5 Hz), 3.5 (1H, dd, 11 and 4.6 Hz), 1.23 (3H, d, 6.75 Hz), 0.83 (3H, s);  $\delta$  ( $\text{C}_6\text{D}_6$ ) 1.00 (3H, d, 6.75 Hz); m/z at 250 ( $\text{M}^+$ , 47); 206 (100); 149 (90)| were identical to those of natural dihydroreynosin<sup>6,10</sup> (m.p. 120°).

Treatment of 7 and 8 with Burgess reagent<sup>11</sup> ( $\text{MeOOC}\bar{\text{N}}\text{SO}_2\text{NEt}_3$ , benzene, 55°C, 3 h) led to enone 12 |m.p. 73°C;  $\nu$  : 1730, 1710, 1690, 1615  $\text{cm}^{-1}$ ;  $\delta$  ( $\text{CDCl}_3$ ) 6.87 (1H, dd, 1.3 Hz, 2.2 Hz), 3.69 (3H, s), 1.3 (3H, s), 1.24 (3H, d, 6.75 Hz); m/z at 264 ( $\text{M}^+$ , 84), 176 (100); UV : 243 nm| in which both carbonyl groups are differentiated because of electronic reasons. Reduction of 12 ( $\text{NaBH}_4$ , MeOH, -30°C, 15 min) followed by catalytic hydrogenation (Pt/C, 5 %, MeOH) and epimerization (DBU,  $\text{CH}_2\text{Cl}_2$ , 3 days, r.t.) afforded 13 in 72 % yield after column chromatography ( $\text{SiO}_2$ , benzene-acetone 8:2) |13; oil;  $\nu$  : 3430, 1730, 1710  $\text{cm}^{-1}$ ,  $\delta$  ( $\text{CDCl}_3$ ) : 3.82 (1H, dd, 4.75 Hz, 11.5 Hz), 3.67 (3H, s), 1.14 (3H, d, 7 Hz), 0.72 (3H, s); m/z at 268 ( $\text{M}^+$ , 8)|. Wittig reaction of 13 with excess ylid (from  $\text{C}_3\text{PCH}_3\text{Br}$  and t.AmONa, toluene, r.t. 30 min) yielded 14 (86 %) after column chromatography ( $\text{SiO}_2$ , ether-hexane 1:1). Formation of the  $\alpha$ -phenylselenide (LDA,  $\text{C}_6\text{H}_5\text{SeSeC}_6\text{H}_5$ , THF, HMPA, -78° to -40°C, 3 h) followed by oxidative elimination ( $\text{H}_2\text{O}_2$ -HOAc-THF, 0°C, 1 h) led to 15 (82 % yield). Collins oxidation of 15 ( $\text{CrO}_3 \cdot 2 \text{Py}$ ,  $\text{CH}_2\text{Cl}_2$ ,  $\text{N}_2$ , r.t.) and column chromatography ( $\text{SiO}_2$ :isooctane-ethylacetate 8:2) afforded ( $\pm$ ) 1-oxocostic acid methyl ester 2 (R = Me) in 83 % yield. The spectral properties of 2 (R = Me) |m.p. 37°C;  $\nu$  : 1710 (broad), 1650, 1625  $\text{cm}^{-1}$ ;  $\delta$  ( $\text{CDCl}_3$ ) : 6.18 (1H, d, 0.5 Hz), 5.58 (1H, t, 1 Hz), 4.99 (1H, d, 1.5 Hz), 4.73 (1H, d, 1.5 Hz), 3.76 (3H, s), 1.01 (3H, s);  $\delta$  ( $\text{C}_6\text{D}_6$ ) : 6.27 (1H, d, 0.5 Hz), 5.35 (1H, t, 12 Hz), 4.84 (1H, d, 1.5 Hz), 4.64 (1H, d, 1.5 Hz), 3.49 (3H, s), 0.87 (3H, s); m/z at 262 ( $\text{M}^+$ , 8); UV : 212 nm| were identical to those of 2 (R = Me) derived from the natural 1-oxocostic acid<sup>7,12</sup>. Hydrolysis ( $\text{K}_2\text{CO}_3$ , MeOH,  $\text{H}_2\text{O}$ , reflux, 12 h) of 2 (R=Me) gave ( $\pm$ ) 1-oxocostic acid 2 (R=H) (83 %), |m.p. 118°C;  $\delta$  ( $\text{CDCl}_3$ ) : 6.34 (1H, br. s), 5.72 (1H, br. s), 5.00 (1H, d, 1.2 Hz), 4.75 (1H, d, 1.2 Hz), 1.03 (3H, s)|.

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REFERENCES

1. Aspirant of the "Nationaal Fonds voor Wetenschappelijk Onderzoek".
2. (a) C.H. Heathcock; The total synthesis of natural products (Ed. J. ApSimon - John Wiley & Sons, 1973) 2, 197; (b) J.W. Huffmann, C.A. Miller, A.R. Pinder; J. Org. Chem., 1976, 41, 3705-3714; (c) W.C. Still, M.J. Schneider; J. Am. Chem. Soc., 1977, 99, 948; (d) J.A. Marschall, G.M. Wutz; J. Org. Chem., 1978, 43, 1086; (e) F. Kido, K. Tsutsumi, R. Maruta, A. Yoshikoshi; J. Am. Chem. Soc., 1979, 101, 6420; (f) B.D. MacKenzie, H.M. Angelo, J. Wolinsky; J. Org. Chem., 1979, 44, 4042; (g) J.D. Godfrey, A.G. Schultz; J. Am. Chem. Soc., 1980, 102, 2414.
3. W.A. Wood; Diss. Abst. Int. B, 1978, 39(2), 760-1.
4. N.H. Fischer, E.J. Olivier, N.H. Fischer; Progress in the Chemistry of Organic Natural Products, 1979, 38, 47.
5. (a) D. Termont, P. De Clercq, D. De Keukeleire, M. Vandewalle; Synthesis, 1977, 46 ; (b) M. Van Audenhove, D. De Keukeleire, M. Vandewalle; Tetrahedron Lett., 1980, 1979; (c) M. Van Audenhove, D. De Keukeleire, M. Vandewalle; Bull. Soc. Chim. Belges, 1981, 90, 255.
6. M. Ogura, G. Cordell and N.R. Farnsworth; Phytochemistry, 1978, 17, 957.
7. F. Bohlmann and J. Jakupovic; Phytochemistry, 1979, 18, 1189.
8. The relative configuration at C<sub>7</sub> and C<sub>10</sub> was proven at the stage of product 7. High stereoselectivity for the photoaddition of piperitone with cyclobutenes has already been observed; see ref. 5c and references cited therein. It is worthwhile mentioning that no stereoselectivity was observed when the cyclohexenone carries a C-6 acetic acid methyl ester side chain.
9. We also observed that methylene triphenylphosphorane reacted exclusively (83 % yield) with the 1-carbonyl group in 9.
10. We thank Dr. G. A. Cordell for kindly sending us copies of the spectra of 1.
11. E.M. Burgess, H.R. Penton Jr. and E.A. Taylor; J. Org. Chem., 1973, 38, 26.
12. We thank Prof. F. Bohlmann for kindly sending us copies of the spectra of 2 (R = Me).

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